A STEREOSPECIFIC APPROACH TO CIS-P-MENTH-2-ENE¹

Anthony Kreft^{*} Department of Chemistry, Columbia University New York, New York 10027

(Received in USA 27 January 1977; received in UK for publication 9 February 1977)

Reaction of lithium organocuprates with allylic esters is a facile method for the formation of new carbon - carbon bonds.² The stereochemistry of this reaction has been recently investigated.³ This data has prompted us to report our relevant findings concerning this reaction in our approach to the synthesis of p-menth-2-enes.



The allylic alcohols⁴, $\underline{18}^5$, $\underline{2a}^{6,7}$, $\underline{3a}^8$ and $\underline{4a}^8$ were first converted (MeLi/Mesitoyl Chloride) to their corresponding mesitoates⁷, <u>1b</u>, <u>2b</u>, <u>3b</u> and <u>4b</u>. Reaction of either <u>1b</u> or <u>3b</u> with 1 equivalent of lithlum dimethylcuprate in ether at 0°C overnight afforded 67% and 84% yields⁹, respectively, of <u>cis-p-menth-2-ene</u>, <u>5</u>. None of the allylic isomers, <u>7</u>, or <u>8</u>, or the epimer <u>6</u>, were present.¹⁰ The structural assignment was further corroborated by reduction of <u>5</u> to <u>9a</u>^{11,12} by dimide. Reaction of either <u>2b</u> or <u>4b</u> with 1 equivalent of lithlum dimethylcuprate in ether at 0°C overnight unfortunately afforded ¹³ in 77% and 57% yields⁹, respectively, the same identical 67:33 mixture of <u>6</u> and <u>8</u>. However, none of the <u>cis-alkenes 5</u> and <u>7</u> could be detected. The structural assignments were corroborated by reduction of the reaction mixture to the saturated hydrocarbons <u>9b</u>¹¹ and <u>10b</u>¹⁴.



Previous approaches¹⁷⁻²² to the <u>p</u>-menth-2-enes have all utilized base-induced elimination reactions of esters, halides or quarternary ammonium salts derived from isomeric menthols. While <u>trans-p</u>-menth-2-ene, <u>6</u>, can be obtained readily from menthol¹⁷⁻²⁰, $\frac{21}{3}$, the <u>cis</u>-isomer, <u>5</u>, is not

easily accessible via this approach. This is due to either inaccessibility of the precursor, isomenthol, 14, or lack or complete stereospecificity in the elimination reaction of the tosylate of neoisomenthol, 15^{22} . Thus, our cuprate approach to <u>p</u>-menth-2-enes is the synthetic complement of the traditional elimination approach.



In addition to the synthetic utility of these results, there are strong mechanistic implications. That the trans-allylic mesitoates, 1b and 3b, gave only the same cis-alkene, 5, and the <u>cis</u> allylic mesitoates, <u>2b</u> and <u>4b</u>, gave the same mixture of <u>trans</u>-alkenes, <u>6</u> and <u>8</u>, concurs with Goerings proposal^{3,23} of a symmetrical intermediate, or its equivalent, which can maintain geometric integrity. However, our results raise new questions about the mechanism of these displacements. Goering's systems, 5-substituted cyclohexenyl substrates, were inherently symmetric with respect to formal SN-2 and SN-2' attack in the proposed intermediate. Our 6-substituted cyclohexenyl substrates clearly show that the relative proportions of formal SN-2 and SN-2' attack can be profoundly influenced by the stereochemistry of an extrinsic asymmetric center. Acknowledgement The author wishes to thank Gilbert Stork and Tim Macdonald for stimulating discussions, and the National Science Foundation and Gilbert Stork for financial support.

References

- Taken in part from the Ph.D. dissertation of Anthony Kreft, Columbia University, 1976. 1.
- G.H. Posner, "Organic Reactions" Vol. 22, John Wiley, New York, 1975. 2.
- H.L. Goering and V.D. Singleton, Jr., J.Amer.Chem.Soc. 98, 7854 (1976). 3.
- 4 All compounds are racemic, although only one enantiomer is depicted.
- G. Stork and W.N. White, J. Amer. Chem. Soc. 78, 4604 (1956). 5.
- Prepared by triisobutylaluminum reduction of the corresponding en-one, mp 43-45. 6.
- All new compounds had satisfactory NMR, IR and MS spectra. 7.
- 8.
- 9.
- G.S. Crishna Raa and S. Dev, J. Ind. Chem. Soc. 33, 539 (1956). Yields were determined by GC using isopropylcyclohexane as an internal standard. The relative GC retention times (10', 1/8'', 5% FFAP, 75°C) of <u>5,6</u> and <u>8</u> were 5.13 min, 10. 4.69 min, and 5.44 min, respectively.
- 11.
- J.F. Sauvage, R.H. Baker and A.S. Hussen, <u>J.Amer.Chem.Soc.</u> 82, 6090 (1960). The relative GC retention times (10' 1/8", 5% FFAP, 75°C) of <u>9a</u>, <u>9b</u>, <u>10a</u> and <u>10b</u> were 12. 4.06 min, 3.50 min, 4.88 min and 4.00 min, respectively.
- It should be noted the corresponding acetate of 2a gave identical results. 13.
- Prepared (a 75:25 mixture of <u>10a</u> and <u>10b</u>) by catalytic reduction¹⁵ of the corresponding 14. exo-methylene compound¹⁶.
- For analogous reduction reduction of the **«**-methyl exomethylene compound: S. Siegal <u>et al.</u>, 15. J.Org.Chem. 31, 2802 (1966).
- B. Willhaus and A.T. Thomas, <u>Helv.Chim.Acta</u> <u>50</u>, 383 (1967)
 C.H. Snyder and H.H. Chang, <u>Chem.Commun.</u>, 114 (1969).
 E.D. Hughes and J. Wilby, <u>J.Chem.Soc.</u>, 4094 (1960). 16.
- 17.
- 18.
- E.D. Hughes, C.K. Ingold and J.B. Rose, J. Chem. Soc., 3839 (1953). 19.
- W. Huckel, W. Tappe and G. Legutke, Justus Liebigs Ann. Chem. 543, 191 (1940). 20.
- W. Huckel and H. Wagner, Chem. Ber. 74, 657 (1941). 21.
- A.L.J. Beckwith and G. Phillipou, Aus.J. Chem. 29, 877 (1976). 22.
- For related work in homoallylic systems see: G.H. Posner, J.S. Ting and C.M. Lentz, Tet. 23. 32, 2281 (1976).

*Present Address: Department of Chemistry, Stanford University, Stanford, Ca. 94305